

Ionic conductivity and crystal structure for the $\text{Li}_{3-2x}\text{Cr}_2-x\text{Ta}_x(\text{PO}_4)_3$ system

Abstract

The monoclinic phase ($P2_1/n$) was formed for $0 \leq x \leq 0.6$ and the NASICON-type rhombohedral phase ($R\bar{3}c$) was obtained for the region $0.8 \leq x \leq 1.2$ in the $\text{Li}_{3-2x}\text{Cr}_2-x\text{Ta}_x(\text{PO}_4)_3$ system. The activation energy for Li^+ migration was ca. 0.45 eV for the monoclinic structure and ca. 0.36 eV for the rhombohedral structure. The maximum conductivity of $8.4 \times 10^{-6} \text{ S cm}^{-1}$ at 298 K was obtained for $x=0.8$ of the $\text{Li}_{3-2x}\text{Cr}_2-x\text{Ta}_x(\text{PO}_4)_3$ system. The conductivity of $\text{LiCrTa}(\text{PO}_4)_3$ was enhanced about three to five times by the addition of the lithium salt due to the improvement of the sinterability. The maximum conductivity was $2.4 \times 10^{-5} \text{ S cm}^{-1}$ at 298 K for $\text{LiCrTa}(\text{PO}_4)_3-0.2\text{Li}_3\text{BO}_3$.

Keywords — Li^+ ionic conductor, solid electrolyte, $\text{LiCrTa}(\text{PO}_4)_3$, NASICON, conductivity.