



**INVENTORS**

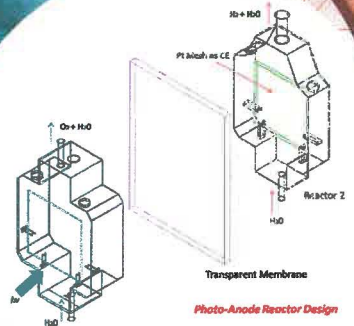
DR. UMI FAZARA MD ALI  
DR MOHD NAZREE DERMAN  
DR MOHD IRFAN HATIM MOHAMED DZAHIR  
TAN TIEK AUN  
SURAINI SULAIMAN

**CONTACT DETAILS**

SCHOOL OF ENVIRONMENTAL ENGINEERING,  
UNIMAP, KOMPLEKS PUSAT PENGAJIAN  
JEAJAWI 3, 02600 ARAU, PERLIS.  
umifazara@unimap.edu.my

# H<sub>2</sub>-PEC-HEMATITE 'REACTORS FOR TOMORROW'S ENERGY': H<sub>2</sub> PRODUCTION VIA PHOTO-ELECTROCHEMICAL REACTOR

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## WHY HYDROGEN?

75% of global electrical energy production involves fossil fuel combustion, resulting in the detrimental effects of CO<sub>2</sub>, SO<sub>x</sub> and NO<sub>x</sub> emissions (Fig. 1). Obviating such effects and fossil fuel depletion are the main drivers for development of energy carriers that are:

- CLEAN
- RENEWABLE
- AVAILABLE

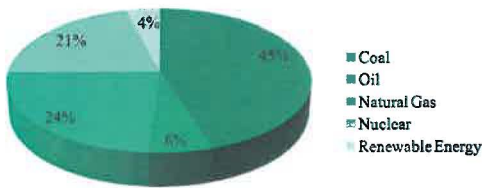


Fig. 1

## MAJOR ADVANTAGES/NOVELTY

- Reactor systems for generation of carbon-free hydrogen using only SUNLIGHT and WATER.
- Reactor behaviour on scale-up for industrial uses, transportation etc.
- Can be used with other metal oxide: Fe<sub>2</sub>O<sub>3</sub>, TiO<sub>2</sub>

## MATERIAL CHALLENGES

Ideally, photo-anodes should:

- Be thermodynamically stable
- Have band gap energy matched the solar spectrum
- Require a potential bias of 1.23 eV
- Cheap and easily available

## INVENTION DETAILS

- Solar radiation is the ultimate source of energy on earth. Semiconductors with band gap energies (E<sub>bg</sub>) matched (ca. 2.5 eV) to the solar spectrum may be used to harvest solar energy.
- (Fig.3) Semiconducting photo-anode absorbs photons with energies,  $h\nu > E_{bg}$ , promoting electrons into semiconductor's conduction band (CB), leaving (electron) 'holes' in its valence band (VB).
- 'Holes' oxidise water:  $2H_2O + 4h_{VB} \rightarrow O_2 + 4H^+$
- Electrons, conducted to metal cathode, reduce water to hydrogen:  $2H_2O + 2e_{CB} \rightarrow H_2 + 2OH^-$

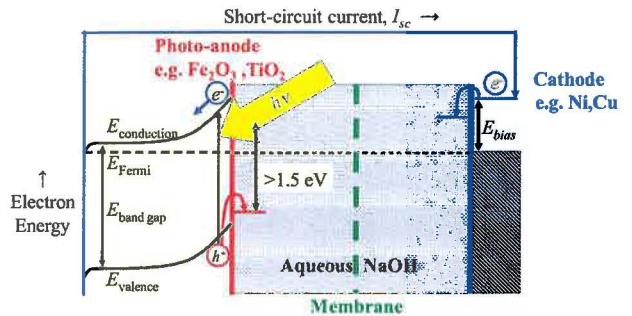


Fig. 2

## ECONOMIC EVALUATION

Current work obtained current density is 6 A/m<sup>2</sup>. According to the Shockley-Quiesser limit, best obtainable performance is 29% for a 1.1 eV bandgap semiconductor which correspond to a current density of 43 mA/cm<sup>2</sup>. The ideal electrode potential to be used is the theoretical limit, 1.23 V vs RHE. 18.61 hectares of space is needed to produce 1000 kg H<sub>2</sub> a day.

**A plant producing 1000kg H<sub>2</sub> a day will cost roughly RM 33560 per day.**

As a fuel 1000kg H<sub>2</sub> can offset 4000 L gasoline which is equivalent to **RM 12960** and offset 9200 kg CO<sub>2</sub> which will cost **RM 29808** to remove.

**SAVING (RM 42768 – RM 33560) = RM 9208 per day using H<sub>2</sub> as fuel**

**PUBLICATIONS**

- A.R., Muhd Hafiz, M.D., Irfan Hatim, M.A. Umi Fazara, A Review: The Development of Malaysia's Alternative Energy Resources, Journal of Renewable & Sustainable Energy (Elsevier), 2013.
- A.R. Muhd Hafiz, T.A. Tan, M.A. Umi Fazara, Fabrication and Characterization of Hematite (Fe<sub>2</sub>O<sub>3</sub>) as Photo-anode Thin Film for Photo-electrochemical Splitting of Water, International Postgraduate on Engineering & Management (IPCEM), UniMAP, 25-26 January 14.
- Muhd Hafiz Abd Rahim, Mohd Irfan Hatim Mohamed Dzhahir, Umi Fazara Md Ali, Development of Hematite (α-Fe<sub>2</sub>O<sub>3</sub>) Photo-Anode Thin Film By Spray Pyrolysis ,9th World Congress of Chemical Engineering (WCCE9), Seoul Korea, 19-23 Aug 13.
- A.R. Muhd Hafiz, T.A. Tan, M.A. Umi Fazara, Characterization and Evaluation of Hematite (Fe<sub>2</sub>O<sub>3</sub>) as Photo-anode Thin Film for Photo-electrochemical Splitting of Water, International Conference of Chemical Engineering (ICCE 2013), San Francisco, USA, 23-25 October 13.